

217475US0PCT

U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR

10/019313

TRANSMITTAL LETTER TO THE UNITED STATES  
DESIGNATED/ELECTED OFFICE (DO/EO/US)  
CONCERNING A FILING UNDER 35 U.S.C. 371INTERNATIONAL APPLICATION NO.  
PCT/EP00/06749INTERNATIONAL FILING DATE  
14 July 2000PRIORITY DATE CLAIMED  
15 July 1999

## TITLE OF INVENTION

## REMOVAL OF MERCAPTANS FROM FLUID STREAMS

APPLICANT(S) FOR DO/EO/US  
HUGO Randolph et al.

Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

- This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
- This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
- This is an express request to begin national examination procedures (35 U.S.C. 371(f)). The submission must include items (5), (6), (9) and (24) indicated below.
- The US has been elected by the expiration of 19 months from the priority date (Article 31).
- A copy of the International Application as filed (35 U.S.C. 371 (c) (2))
  - is attached hereto (required only if not communicated by the International Bureau).
  - has been communicated by the International Bureau.
  - is not required, as the application was filed in the United States Receiving Office (RO/US).
- An English language translation of the International Application as filed (35 U.S.C. 371(c)(2))
  - is attached hereto.
  - has been previously submitted under 35 U.S.C. 154(d)(4).
- Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371 (c)(3))
  - are attached hereto (required only if not communicated by the International Bureau).
  - have been communicated by the International Bureau.
  - have not been made; however, the time limit for making such amendments has NOT expired.
  - have not been made and will not be made.
- An English language translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
- An oath or declaration of the inventor(s) (35 U.S.C. 371 (c)(4)).
- An English language translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371 (c)(5)).
- A copy of the International Preliminary Examination Report (PCT/IPEA/409).
- A copy of the International Search Report (PCT/ISA/210).

## Items 13 to 20 below concern document(s) or information included:

- An Information Disclosure Statement under 37 CFR 1.97 and 1.98.
- An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
- A **FIRST** preliminary amendment.
- A **SECOND** or **SUBSEQUENT** preliminary amendment.
- A substitute specification.
- A change of power of attorney and/or address letter.
- A computer-readable form of the sequence listing in accordance with PCT Rule 13ter.2 and 35 U.S.C. 1.821 - 1.825.
- A second copy of the published international application under 35 U.S.C. 154(d)(4).
- A second copy of the English language translation of the international application under 35 U.S.C. 154(d)(4).
- Certificate of Mailing by Express Mail
- Other items or information:

PCT/IB/304/Drawings (1 sheet)

PCT/IB/308/Notice of Priority/Form PTO-1449

Amended Sheets (Pages 16 and 17)

|  |   |   |
|--|---|---|
| U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR<br><b>10/019313</b> | INTERNATIONAL APPLICATION NO.<br>PCT/EP00/06749 | ATTORNEY'S DOCKET NUMBER<br>21475US0PCT |
|--|---|---|

24. The following fees are submitted.

**BASIC NATIONAL FEE (37 CFR 1.492 (a) (1) - (5)) :**

|  |           |
|--|-----------|
| <input type="checkbox"/> Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO and International Search Report not prepared by the EPO or JPO ..... | \$1040.00 |
| <input checked="" type="checkbox"/> International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO .....  | \$890.00  |
| <input type="checkbox"/> International preliminary examination fee (37 CFR 1.482) not paid to USPTO but international search fee (37 CFR 1.445(a)(2)) paid to USPTO .....  | \$740.00  |
| <input type="checkbox"/> International preliminary examination fee (37 CFR 1.482) paid to USPTO but all claims did not satisfy provisions of PCT Article 33(1)-(4) .....   | \$710.00  |
| <input type="checkbox"/> International preliminary examination fee (37 CFR 1.482) paid to USPTO and all claims satisfied provisions of PCT Article 33(1)-(4) .....   | \$100.00  |

**CALCULATIONS PTO USE ONLY**

**ENTER APPROPRIATE BASIC FEE AMOUNT = \$890.00**

Surcharge of **\$130.00** for furnishing the oath or declaration later than  20  30 months from the earliest claimed priority date (37 CFR 1.492 (e)). **\$0.00**

| CLAIMS   | NUMBER FILED | NUMBER EXTRA | RATE                                   |
|--|--------------|--------------|--|
| Total claims                                     | 10 - 20 =    | 0            | x \$18.00 <b>\$0.00</b>                |
| Independent claims                               | 2 - 3 =      | 0            | x \$84.00 <b>\$0.00</b>                |
| Multiple Dependent Claims (check if applicable). |              |              | <input type="checkbox"/> <b>\$0.00</b> |

**TOTAL OF ABOVE CALCULATIONS = \$890.00**

Applicant claims small entity status. See 37 CFR 1.27). The fees indicated above are reduced by 1/2. **\$0.00**

**SUBTOTAL = \$890.00**

Processing fee of **\$130.00** for furnishing the English translation later than  20  30 months from the earliest claimed priority date (37 CFR 1.492 (f)). **\$0.00**

**TOTAL NATIONAL FEE = \$890.00**

Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31) (check if applicable). **\$0.00**

**TOTAL FEES ENCLOSED = \$890.00**

|                                      |    |
|--------------------------------------|----|
| Amount to be:<br>refunded<br>charged | \$ |
|                                      |    |

- A check in the amount of **\$890.00** to cover the above fees is enclosed.
- Please charge my Deposit Account No. \_\_\_\_\_ in the amount of \_\_\_\_\_ to cover the above fees. A duplicate copy of this sheet is enclosed.
- The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. **15-0030** A duplicate copy of this sheet is enclosed.
- Fees are to be charged to a credit card. **WARNING:** Information on this form may become public. Credit card information should not be included on this form. Provide credit card information and authorization on PTO-2038.

NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.

SEND ALL CORRESPONDENCE TO:

Surinder Sachar  
Registration No. 34,423



**22850**

*Surinder Sachar*  
SIGNATURE

Norman F. Oblon

NAME

24,618

REGISTRATION NUMBER

*Jan. 9 2002*  
DATE

217475US-0 PCT

IN THE UNITED STATES PATENT & TRADEMARK OFFICE

IN RE APPLICATION OF:

RANDOLF HUGO ET AL

: ATTN: APPLICATION DIVISION

SERIAL NO: NEW U.S. PCT APPLN

(BASED ON PCT/EP00/06749)

FILED: HEREWITH

:

FOR: REMOVAL OF MERCAPTANS  
FROM FLUID STREAMSPRELIMINARY AMENDMENTASSISTANT COMMISSIONER FOR PATENTS  
WASHINGTON, D.C. 20231

SIR:

Prior to examination on the merits, please amend the above-identified application as follows.

IN THE CLAIMS

Please delete the title and replace with the following title:

METHOD FOR REMOVING MERCAPTANS FROM FLUID FLUXES

IN THE CLAIMS

Please amend the claims as shown on the marked-up copy following this amendment to read as follows.

3. (Amended) A process as claimed in claim 1, wherein the scrubbing liquor contains not more than 5% by weight of a physical solvent for mercaptans.
4. (Amended) A process as claimed in claim 1, wherein the tertiary alkanolamine used is methyldiethanolamine.

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5. (Amended) A process as claimed in claim 1, wherein the scrubbing liquor contains as activator, a primary or secondary alkanolamine or a saturated 5- or 6-membered N-heterocycle which optionally contains further heteroatoms selected from oxygen and nitrogen.

9. (Amended) The use as claimed in claim 7, wherein the activator is piperazine or methylpiperazine.

REMARKS

Claims 1-10 are active in the present application. Claims 3-5 and 9 have been amended to remove multiple dependency. No new matter is added. An action on the merits and allowance of claims is solicited.

Respectfully submitted,

OBLON, SPIVAK, McCLELLAND,  
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**Marked-Up Copy**

Serial No: \_\_\_\_\_

Amendment Filed on: \_\_\_\_\_

**IN THE TITLE**

Please delete the title and replace with the following title.

[METHOD FOR REMOVING MERCAPTANS FROM FLUID FLUXES]

--REMOVAL OF MERCAPTANS FROM FLUID STREAMS--

**IN THE CLAIMS**

--3. (Amended) A process as claimed in claim 1 [or 2], wherein the scrubbing liquor contains not more than 5% by weight of a physical solvent for mercaptans.

4. (Amended) A process as claimed in [any of claims 1 to 3] claim 1, wherein the tertiary alkanolamine used is methyldiethanolamine.

5. (Amended) A process as claimed in [any of claims 1 to 4] claim 1, wherein the scrubbing liquor contains as activator, a primary or secondary alkanolamine or a saturated 5- or 6-membered N-heterocycle which optionally contains further heteroatoms selected from oxygen and nitrogen.

9. (Amended) The use as claimed in [either of claims 7 and 8] claim 7, wherein the activator is piperazine or methylpiperazine.--

1/Prb  
Removal of mercaptans from fluid streams

The present invention relates to a process for removing 5 mercaptans from fluid streams comprising same, especially from hydrocarbon gas streams, for example natural gas, synthesis gas from heavy oil or heavy residues or refinery gas, or else from liquid hydrocarbons, for example LPG (liquefied petroleum gas).

10 Numerous processes in the chemical industry give rise to fluid streams comprising acid gases, for example  $\text{CO}_2$ ,  $\text{H}_2\text{S}$ ,  $\text{SO}_2$ ,  $\text{CS}_2$ ,  $\text{HCN}$ ,  $\text{COS}$  or mercaptans as impurities.

The LPG or gas streams in question here can be for example 15 hydrocarbon gases from a natural gas source, synthesis gases from chemical processes or, say, reaction gases involved in the partial oxidation of organic materials, for example coal or petroleum. The removal of sulfur compounds from these fluid streams is of particular importance for various reasons. For 20 instance, the level of sulfur compounds in natural gas has to be reduced by suitable processing measures immediately at a natural gas well, since the natural gas will normally also contain a certain fraction of entrained water as well as the above-recited sulfur compounds. In aqueous solution, however, these sulfur 25 compounds form acids and have a corrosive effect. To transport natural gas in a pipeline, therefore, predetermined limits must be complied with for the sulfur-containing impurities. In addition, numerous sulfur compounds are malodorous and - with hydrogen sulfide ( $\text{H}_2\text{S}$ ) a prime example - extremely toxic even at 30 low concentrations.

Similarly, the  $\text{CO}_2$  content of hydrocarbon gases, such as natural gas, customarily has to be significantly reduced, since high concentrations of  $\text{CO}_2$  reduce the calorific value of the gas and 35 may likewise cause corrosion to pipework and fittings.

There are therefore numerous processes already in existence for removing acid gas constituents from fluid streams such as hydrocarbon gases or LPG. In the most widely used processes, the 40 fluid mixture containing acid gases is contacted with an organic solvent or an aqueous solution of an organic solvent as part of a gas scrub process.

There is extensive patent literature on gas scrub processes and the scrubbing solutions used in these processes. In principle, two different kinds of gas scrub solvents can be distinguished:

5 On the one hand there are physical solvents, which rely on a physical absorption process, i.e., the acid gases dissolve in the physical solvent. Typical physical solvents are cyclotetramethylene sulfone (sulfolane) and its derivatives, 10 aliphatic acid amides, NMP (N-methylpyrrolidone), N-alkylated pyrrolidones and corresponding piperidones, methanol and mixtures of dialkylethers of polyethylene glycols (Selexol®, Union Carbide, Danbury, Conn., USA).

15 On the other hand, there are chemical solvents which work on the basis of chemical reactions which convert the acid gases into compounds which are simpler to remove. For instance, the most widely used chemical solvents in industry, aqueous solutions of alkanolamines, form salts when acid gases are passed through, and 20 these salts can either be decomposed by heating and/or stripped off by means of steam. The alkanolamine solution is regenerated in the course of the heating or stripping, so that it can be re-used. Preferred alkanolamines used for removing acid gas impurities from hydrocarbon gas streams include monoethanolamine 25 (MEA), diethanolamine (DEA), triethanolamine (TEA), diisopropylamine (DIPA), diglycolamine (DGA) and methylidiethanolamine (MDEA).

Primary and secondary alkanolamines are particularly suitable for 30 gas scrubs where the purified gas has to have a very low CO<sub>2</sub> content (e.g., 10 ppmv of CO<sub>2</sub>). To remove H<sub>2</sub>S from gas mixtures having a high original CO<sub>2</sub> content, however, it has been found to be disadvantageous that the effectiveness of the solution for removing H<sub>2</sub>S is much reduced by an accelerated absorption of CO<sub>2</sub>. 35 In addition, the regeneration of solutions of primary and secondary alkanolamines consumes large volumes of steam.

The European patent application EP-A-0 322 924 discloses, for example, that tertiary alkanolamines, especially MDEA, are 40 particularly suitable for a selective removal of H<sub>2</sub>S from gas mixtures containing H<sub>2</sub>S and CO<sub>2</sub>.

The German patent application DE-A-1 542 415 proposes increasing the effectiveness not only of physical solvents but also of 45 chemical solvents by addition of monoalkylalkanolamines or of morpholine and its derivatives. The German patent application DE-A-1 904 428 describes the addition of monomethylethanolamine

(MMEA) as an accelerator to improve the absorption properties of an MDEA solution.

U.S. Patent 4,336,233 describes one of the currently most effective scrubbing solutions for removing CO<sub>2</sub> and H<sub>2</sub>S from a gas stream. It is an aqueous solution of about 1.5 to 4.5 mol/l of methyldiethanolamine (MDEA) and 0.05 to 0.8 mol/l of piperazine as absorption accelerant (aMDEA®, BASF AG, Ludwigshafen). The removal of CO<sub>2</sub> and H<sub>2</sub>S using MDEA is further described in greater detail in the following patents of present assignee: US 4,551,158; US 4,553,984; US 4,537,753; US 4,999,031, CA 1 291 321 and CA 1 295 810. The removal of mercaptans from gas streams containing same is not mentioned in these property rights.

15 Mercaptans are substituted forms of H<sub>2</sub>S in which a hydrocarbyl R takes the place of one of the hydrogen atoms. Their general formula is therefore RSH. The properties of mercaptans depend substantially on the length of the hydrocarbon chain. Mercaptans in aqueous solution likewise act as acids, but are significantly weaker than H<sub>2</sub>S, for example. With increasing length of the hydrocarbon chain, therefore, mercaptans behave like hydrocarbons, which makes their removal from hydrocarbon gas streams particularly difficult. For instance, there is a report in the literature thatMEA and DEA solutions will remove about 45 to 50% of methyl mercaptan, but only 20 to 25% of ethyl mercaptan and about 0 to 10% of propyl mercaptan (A. Kohl, R. Nielsen: "Gas Purification", 5th Edition, 1997, p.155). "Gas Conditioning and Processing", Vol. 4: "Gas Treating and Liquid Sweetening", 4th Ed., J.M. Campbell & Company, 1998, states on page 51 that aqueous 20 amine solutions have little if any utility with regard to the removal of mercaptans from gas streams. Mercaptans occur in some natural gas sources, especially on the North American continent, and are typically present in most liquid or liquefied refined hydrocarbon products (LPG). However, because of their corrosive 25 and malodorous properties, mercaptans must likewise be substantially removed from hydrocarbon gases or liquids. Treated and purified hydrocarbons for polymerization reactions, for example, should customarily contain not more than 1 - 20 ppm of mercaptans.

30 The literature contains a wide variety of proposals for removing mercaptans from fluid streams containing same:

U.S. Patent 4,808,765 describes a three-stage process for 45 removing acid gases from a gaseous hydrocarbon stream. The first step is an absorption process in which an aqueous solvent which contains MDEA as a selective absorbent for H<sub>2</sub>S and DIPA as a

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selective absorbent for COS, to remove substantially all the H<sub>2</sub>S and a portion of the COS. The second step, which utilizes an aqueous alkaline solution of a primary alkanolamine as scrubbing solution, removes a substantial portion of the remaining COS. The 5 third step, finally, removes mercaptans with the aid of an aqueous caustic solution (NaOH). This process is very complicated in terms of apparatus, since the individual scrubbing solutions have to be regenerated separately. Moreover, the hydrocarbon gas stream has to be subsequently additionally scrubbed with water to 10 remove remnants of the caustic solution.

U.S. Patent 4,462,968 states that, although traditional alkanolamine solutions are capable of removing H<sub>2</sub>S down to concentrations of less than 4 ppm, these processes are not 15 suitable for removing mercaptans. US 4,462,968 therefore proposes a scrubbing solution for the removal of mercaptans which consists of hydrogen peroxide or a combination of hydrogen peroxide with ammonia or with an amine. However, this process can be operated as a one-step process only with regard to gas streams having a 20 sulfur content of not more than 50 ppm. At a higher sulfur content, it is necessary to operate a two-step process in which the first step involves using an alkanolamine scrubbing solution to remove H<sub>2</sub>S and the second step employs a hydrogen peroxide scrubbing solution to remove mercaptans, sulfides and disulfides.

25 U.S. Patent 4,484,934 describes neat methoxyethylpyrrolidone as a physical solvent for removing mercaptans and other sulfur compounds from a gas stream. It further describes a solvent consisting of water, amine and methoxyethylpyrrolidone.

30 Lastly, the international patent application WO 95/13128 describes a process and a solvent for absorbing mercaptans from gas streams, the scrubbing solution comprising a polyalkylene glycol alkyl ether, for example methoxytriglycol, a secondary 35 monoalkanolamine and optionally further amines, such as MDEA or DEA.

However, the use of a physical solvent such as methoxytriglycol for removing mercaptans from gas streams is associated with 40 disadvantages. Physical solvents are typically used in excess, so that not only mercaptans but also a large fraction of product of value, i.e., hydrocarbon gases in the case of natural gas, are absorbed in the solvent. The increasing absorption of hydrocarbons with increasing pressure is disadvantageous in a 45 high pressure natural gas scrub in particular. This is because the absorbed product of value is then either burned as flash gas and accordingly lost or recycled into the absorber feed, which,

because of the recompression required and on account of the increase in the internal stream, leads to an increase in the size of the plant and to higher operating costs.

5 It is an object of the present invention to provide a simple and economical process for reliably removing mercaptans as well as other acid gas constituents from gaseous or liquid hydrocarbon streams.

10 We have found that this object is achieved by the process of the present claim 1. The invention accordingly provides a process for removing mercaptans from a fluid stream comprising mercaptans and further acid gases, especially CO<sub>2</sub> and/or H<sub>2</sub>S, which comprises intimately contacting the fluid stream in an absorption or

15 extraction zone with a scrubbing liquor comprising at least one aliphatic alkanolamine of 2 - 12 carbon atoms, the amount of scrubbing liquor being supplied to the absorption or extraction zone being sufficient to remove at least CO<sub>2</sub> and H<sub>2</sub>S essentially completely from the fluid stream. The intimate contact between

20 fluid stream and scrubbing liquor in the absorption zone ensures that mercaptans and other acid gases are absorbed by the scrubbing liquor. The substantially decontaminated lean fluid stream and the loaden scrubbing liquor are then separated and discharged from the absorption or extraction zone. The loaden 25 scrubbing liquor contaminated with mercaptans and other acid gas constituents and discharged from the absorption zone is then customarily regenerated. The regenerated lean scrubbing liquor can then be recycled back into the absorption zone.

30 The fluid stream of the process of the invention can be a gaseous or liquid hydrocarbon stream. Natural gas is a typical example of a gas stream, while LPG is an example of a liquid stream.

In the process of the invention, the scrubbing liquor is 35 preferably an aqueous solution and contains from 10 to 70% by weight of the aliphatic alkanolamine with particular advantage. Any reference in the present context to an aliphatic alkanolamine is also to be understood as encompassing a mixture of different alkanolamines, in which case the above-stated percentages then 40 relate to the total alkanolamine content.

The process of the invention is distinguished from existing processes for removing mercaptans from fluid streams in that the scrubbing liquor used contains only a small fraction, preferably 45 not more than 5% by weight, of a physical solvent for mercaptans. It is particularly preferable for the scrubbing liquor not to contain any of the customary physical solvents for mercaptans.

While mercaptans and other acid gases possess a certain solubility even in water and in alkanolamines, these are not deemed to be physical solvents in the proper sense. On the contrary, physical solvents for the purposes of the present invention are in particular those typical physical solvents used in gas scrubbing, such as cyclotetramethylene sulfone (sulfolane), aliphatic acid amides, NMP, N-alkylated pyrrolidones, methanol or alkyl or dialkyl ethers of polyethylene glycol. Such solvents are preferably not employed in the scrubbing liquor of the invention, since the excess operation envisaged according to the invention would lead to a high loss of the product of value, the hydrocarbon gas.

Aqueous alkanolamine solutions have hitherto merely been used for removing  $H_2S$  and  $CO_2$ . The process of the invention surprisingly makes it possible to use these scrubbing liquors, which are known per se, to remove mercaptans from fluid streams, too. The aspect which must be a particular surprise to those skilled in the art is the observation underlying the invention that it is sufficient to dimension an absorption column in such a way that any  $CO_2$  present in the feed gas and any  $H_2S$  present in the feed gas are essentially completely removed from the fluid stream. The amount of scrubbing liquor this requires then automatically leads to a substantial removal of mercaptans from the fluid stream. For example, the process of the invention provides a reduction in the mercaptan content of natural gas by from 75% to 95%, which is simply considered impossible in the literature for an amine scrub, i.e., a scrub with an aqueous amine solution as scrubbing liquor.

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The removal of  $CO_2$  and  $H_2S$  from a hydrocarbon fluid stream is familiar to those skilled in the art. There is already commercial software available which, on the basis of predetermined plant parameters and the specifications desired for the purified gas or LPG, can calculate the operating parameters for a certain scrubbing liquor (an example is the TSWEET program from Brian Research & Engineering). The invention proposes dimensioning the operating parameters in such a way that the  $CO_2$  and  $H_2S$  levels in a given fluid stream are lowered for example to not more than 500 ppm, preferably 50 ppm of  $CO_2$  and not more than 10 ppm, preferably 4 ppm of  $H_2S$ , respectively. The required amount of scrubbing liquor which can be calculated on that basis will according to the invention also remove a very large portion of the mercaptans present in the fluid stream.

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The process of the invention provides for substantial removal of mercaptans from the fluid stream while at the same time only relatively small amounts of gaseous or (in the case of LPG) liquid hydrocarbons are dissolved in the scrubbing liquor. There is thus hardly any loss of product of value, and the disadvantages of the physical solvents traditionally used for mercaptan removal are avoided. Typically, the scrubbing liquor discharged from the absorption region contains less than 1% by weight of hydrocarbons, preferably less than 0.3% by weight of hydrocarbons, particularly preferably less than 0.1% by weight of hydrocarbons.

The inventors determined that effective, i.e., substantial, mercaptan removal (i.e. essentially methyl mercaptan, ethyl mercaptan and propyl mercaptan) requires the removal of all major acid gas components (for example, in the case of a natural gas stream, mainly CO<sub>2</sub>, H<sub>2</sub>S, COS). It is not possible to substantially remove mercaptans while, for example, CO<sub>2</sub> or H<sub>2</sub>S are only removed incompletely and are still present in the treated gas in the 15 percent range, say. The absorption of the individual components takes place roughly in the order of the acid strength, i.e., essentially in the order of H<sub>2</sub>S, CO<sub>2</sub>, COS, mercaptans. Since the mercaptans, as very weak acids, are absorbed by the scrubbing liquor as the last component, as it were, the invention proposes 20 offering an excess of scrubbing liquor in order that the mercaptans may be absorbed as well as H<sub>2</sub>S, CO<sub>2</sub> and COS. The inventors determined that an insufficient amount of solvent leads to a displacement of the mercaptans by the stronger acids, with the result that only little mercaptan is absorbed. Typical values 25 as chosen according to the invention for natural gas scrubbing, for example, are frequently within the range from 10 to 50 liters of scrubbing liquor per cubic meter (s.t.p.) of acid gas in the gas stream (m<sup>3</sup> s.t.p. = m<sup>3</sup> at 0 °C and 101.325 kPa (1.01325 bar absolute)). However, it is impossible to define the excess 30 precisely, since the absorption of acid gas constituents in the scrubbing liquor proposed by the invention is not precisely stoichiometric. More particularly, the optimum ratio of scrubbing liquor to the acid gas fraction in the feed gas or feed LPG will depend on the equilibrium conditions which in turn depend on the 35 respective operating parameters, in the case of a gas scrub in particular on the feed gas temperature and the feed gas pressure, the feed gas composition, the temperature of the (regenerated) scrubbing liquor, the residual contamination of the scrubbing liquor, the absorber base-of-column temperature, the separating 40 efficiency of the column (number of plates or height equivalent to a theoretical plate), etc., although the absorber base-of-column temperature is usually not a free parameter, but 45

is determined by the heat of absorption. On the basis of the fundamentals described, a person skilled in the art is able to compute the requisite excess of scrubbing liquor for the particular operating conditions using, for example, the 5 abovementioned TSWEET program and optimize the operating conditions in actual service, starting from the computed values, by means of a few series of experiments.

It is true that there are at present no commercial programs for 10 mercaptan removal using an amine scrub, since amine scrubs were hitherto considered unsuitable for this purpose. With the process proposed by the invention, however, mercaptan removal can be based on the removal of the acid gases CO<sub>2</sub> and H<sub>2</sub>S which are traditionally removable using an amine scrub. For example, using 15 the TSWEET program, it is possible to compute a scrubbing liquor quantity to provide for 95% removal of CO<sub>2</sub> and H<sub>2</sub>S. The invention then provides that this theoretically determined solvent quantity be raised by from 5 to 30%, preferably by from 10 to 20%. This excess, then, would then also provide for the removal of a very 20 large portion of the mercaptans present in the fluid stream.

The aliphatic alkanolamine used is preferably a tertiary alkanolamine, for example triethanolamine (TEA) or 25 methyldiethanolamine (MDEA), the use of MDEA being particularly preferred for gas streams.

The scrubbing liquor advantageously further contains from 0 to 20% by weight of a primary or secondary amine as activator, especially of a primary or secondary alkanolamine or of a 30 saturated 5- or 6-membered N-heterocycle which optionally contains further heteroatoms selected from O and N. The activator is advantageously selected from the group consisting of monoethanolamine, monomethylethanolamine, diethanolamine, piperazine, methylpiperazine and morpholine. The preferred 35 activator used in the process of the invention is piperazine in a concentration of from 0.5 to 15% by weight, particularly preferably from 3 to 8% by weight.

The process of the invention can be carried out with the 40 customary scrubbing means used in gas scrubbing or LPG scrubbing. Suitable scrubbing means, which ensure an intimate contact between the fluid stream and the scrubbing liquor, are for example randomly packed, structurally packed and plate columns, radial flow scrubbers, jet scrubbers, venturi scrubbers and 45 rotational spray scrubbers, preferably structurally packed, randomly packed and plate columns.

The temperature of the scrubbing liquor in the absorption column is typically within the range from 40 to 70°C at the top of the column and from 50 to 100°C at the base of the column. The overall pressure in the column is generally within the range from 1 to 5 120 bar, preferably within the range from 10 to 100 bar.

The process of the invention can be carried out in one step or in a plurality of successive substeps. In the latter case, the fluid stream containing the acidic gas constituents is intimately 10 contacted in each substep with a separate substream of the scrubbing liquor. For example, various locations in the absorption zone can be supplied with a substream in the absorbent, in which case - if an absorption column is used, for example - the temperature of the supplied scrubbing liquor in 15 successive substeps generally decreases from the base to the top of the column.

The scrubbing liquor contaminated with acidic gas constituents can be regenerated and subsequently returned into the absorption 20 zone with reduced contamination. Typically, in the course of the regeneration, the contaminated scrubbing liquor is decompressed from a relatively high pressure, prevailing in the absorption zone, to a lower pressure. Decompression can be accomplished by means of a throttle valve, for example. Additionally or 25 alternatively, the scrubbing liquor can be passed through an expansion turbine with which a generator may be driven and electric energy may be obtained. The energy thus removed from the scrubbing liquor in the course of expansion can be also used, for example, to drive liquid pumps in the scrubbing liquor 30 recirculation system.

The removal of the acidic gas constituents to regenerate the scrubbing liquor can be effected, for example, in an expansion column, for example a vertical or horizontal flash vessel or a 35 countercurrent column fitted with internals. There may be a plurality of consecutive expansion columns in which regeneration is effected at different pressures. For example, the scrubbing liquid can be initially regenerated in a pre-expansion column at high pressure, for example at about 1.5 bar above the partial 40 pressure of the acidic gas constituents in the absorption zone, and then in a main expansion column at low pressure, for example at from 1 to 2 bar absolute. If a multistage expansion process is used, the first expansion column preferably removes inert gases, such as absorbed hydrocarbons, and the subsequent expansion 45 columns, the acidic gas constituents.

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Preferably, the scrubbing liquor to be regenerated is also subjected to a stripping process to remove further acid gases. To this end, the scrubbing liquor and a stripping agent, advantageously a hot gas (steam is preferred), is passed 5 countercurrently through a desorption column equipped with random packings, structured packings or plates. Preferably, the stripping pressure is from 1 to 3 bar absolute at a temperature from 90 to 130°C.

10 A regeneration of the scrubbing liquor in a plurality of successive substeps in which the contamination of the scrubbing liquor with acid gas constituents decreases with every substep is described in, for example, US 4,336,233, where a coarse scrub is carried out with an expansion cycle only and no stripping, and 15 the contaminated scrubbing liquor is decompressed through an expansion turbine and regenerated stepwise in a pre-expansion column and a main expansion column. This variant is used in particular when the acidic gases to be scrubbed out have high partial pressures and when the clean gas has to meet only low 20 purity requirements.

In a further preferred embodiment of the process of the present invention, the scrubbing liquor substreams used in successive substeps of the scrubbing or absorption process are obtainable 25 through successive substeps of the regeneration process and have a decreasing contamination with acidic gas constituents. In a particularly preferred process, the feed gas or LPG containing the acidic constituents are intimately contacted in succession with a first substream of the scrubbing liquor (obtained after 30 partial regeneration in an expansion column and prior to stripping) and a second substream of the scrubbing liquor (obtained after stripping).

For example, as described in US 4,336,233, the absorption step 35 can be carried out in two substeps, a coarse scrub and a fine scrub, and the regeneration step stepwise through decompression in an expansion turbine, a pre-expansion column and a main expansion column, and also through subsequent stripping. In this case, the substream of the scrubbing liquor for the coarse scrub 40 can come from the main expansion column and the substream for the fine scrub from the stripping stage.

The regenerated absorbent, before it is introduced into the absorption zone, is customarily passed through a heat exchanger 45 to adjust it to the temperature required for the scrub. For example, the regenerated scrubbing liquor leaving the stripping column can have heat removed from it and supplied to the

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scrubbing liquor still containing acid gas constituents prior to its entry into the stripping column.

The process of the invention can be carried out using typical 5 plant configurations used for gas scrubbing and subsequent regeneration of the scrubbing liquor, as described for example in US 4,336,233 for a one-stage or two-stage scrubbing process and particularly extensively in EP-A 0 322 924 for a single-stage scrubbing process featuring an expansion and stripping step. The 10 two documents are hereby expressly incorporated herein by reference.

The invention further proposes that conventional activated aqueous methyldiethanolamine solutions, hitherto merely used for 15 removing CO<sub>2</sub> and H<sub>2</sub>S from gas streams, also be used for removing mercaptans from fluid streams containing same. The present invention accordingly also provides for the use of an activated aqueous MDEA solution for removing mercaptans from fluid streams containing same, especially from hydrocarbon gases such as 20 natural gas or from LPG. Such scrubbing liquors are being marketed as highly concentrated solutions, for example under the brand name of AMDEA® (manufacturer: BASF AG, Ludwigshafen, Germany) with piperazine as activator. The user dilutes the highly concentrated solution with water until the solution has 25 approximately the following composition: from 10 to 70% by weight of methyldiethanolamine, from 0.5 to 15% by weight of piperazine and from 30 to 60% by weight of water.

The process of the invention will now be more particularly 30 described with reference to the accompanying drawing. The figure of the drawing illustrates a use example of the process of the invention in a single-stage scrubbing process followed by a regeneration of the scrubbing liquor using expansion and stripping columns.

35 Referring to the figure, a preferred arrangement can be seen for carrying out the process of the invention, as used for example for removing mercaptans from a natural gas stream containing mercaptans and further acid gases.

40 The fluid mixture, which may for example contain natural gas as product of value and in addition acid gases such as H<sub>2</sub>S, CO<sub>2</sub> and mercaptans, is passed through a feed line 10 into an absorption column 11. Upstream of the point of entry into the absorption 45 column there may be provided separating means (not depicted), for example to remove droplets of liquid from the crude gas. The absorption column 11 possesses an absorption zone 12 which

ensures intimate contact between the acidic crude gas and a scrubbing liquor which is lean with regard to acid gases and which passes into the head region of the absorption column 11 via feed line 13 and is passed countercurrently to the gas to be treated. The absorption region 12 can be realized for example in the form of plates, for example sieve or bubble cap plates, or through packing. Typically, the number of plates used is from 20 to 34. In the head region of the absorption column 11 there may be disposed from 1 to 5 backwash plates 14 so as to reduce the loss of volatile constituents of the scrubbing liquor. The backwash plates 14, constructed as bubble cap plates, for example, are fed via condensate line 15 with water through which the treated gas is passed.

15 The natural gas stream substantially freed of acid gas constituents including the mercaptans leaves the absorption column 11 via a head takeoff 16. The line 16, especially if no backwash plates are provided in the column 11, may be equipped on the inside with a separator (not depicted) to remove entrained 20 scrubbing liquor from the gas stream.

Instead of the single-stage absorption facility described herein, it is also possible to use a two-stage variant as depicted for example in Figure 2 of U.S. patent 4,336,233.

25 The scrubbing liquor containing acid gas leaves the absorption column 11 via line 17 and passes through an optional expansion turbine 18 and a line 19 into the head region of a first expansion column 20. In the expansion column 20, the scrubbing 30 liquor pressure is suddenly decreased, so that the lighter hydrocarbons are able to evaporate from the scrubbing liquor. These hydrocarbons can be incinerated or returned into absorption column 11. However, the process of the invention is notable for the fact that the fraction of absorbed hydrocarbons in the 35 scrubbing liquor leaving the absorption column 11 is very low - despite the excess supply of scrubbing liquor. It is therefore usually possible to dispense with complicated recycling of the hydrocarbons from the expansion column 20 into the absorption column 11. The scrubbing liquor leaves the first expansion column 40 20 via line 21 at the base of the column, whereas the evaporated hydrocarbons are withdrawn via line 21 at the top of the expansion column 20.

In the depicted example, the scrubbing liquor then passes into a 45 second expansion column 23, which can be configured as a low pressure flash column, for example. Less volatile acid gases evaporate off via the line 25, having passed through optional

backwash plates 24. At the top of the second expansion column 23 there can be provided a heat exchanger with head distributor or condenser 26 to return entrained droplets of the scrubbing liquor into the expansion column. The condenser 26 can optionally be 5 bypassed by a bypass line 27. The scrubbing liquor leaves the second expansion column 23 via line 28 and is pumped via pump 29 through a heat exchanger 30, where it takes up heat from the regenerated scrubbing liquor returned to the absorption column 11. The line 28 can be equipped on the inside with an 10 intermediate vessel 31 in which the COS is hydrolyzed, if desired. The scrubbing liquor then passes into the head region of a stripping column 32 in which the scrubbing liquor passes countercurrently to a gas stream, for example steam. The stripping column 32 removes remaining acid gas constituents from 15 the scrubbing liquor. The scrubbing liquor leaves the base region of the stripping column 32 via line 33, while the stripped-off acid gas constituents return via line 34 into the base region of the second expansion column 23. The scrubbing liquor leaving through the line 33 passes to a distributor 35 where a portion of 20 the scrubbing liquor is transported via line 36 to a reboiler 38, which heats the liquid and returns it as vapor via line 39 into the stripping tube. Another portion of the scrubbing liquor passes from the distributor 35 via line 37 to a pump 40, which, as shown schematically in the form of the transfer path 41, is 25 connected to the expansion turbine 18. The expansion turbine supplies a portion of the energy required to drive the pump 40. The regenerated scrubbing liquor, which is lean in acid gases, passes via line 42 into the heat exchanger 30, where it transfers heat to the scrubbing liquor passing via line 28 into the 30 stripping column 32. The regenerated scrubbing liquor then passes through the lines 43 and 13 back into the absorption column 11 to again absorb acid gases. Upstream of the point of entry into the absorption column there can be provided a further heat exchanger 44 to cool the scrubbing liquor to the requisite feed 35 temperature. Similarly, filters and other cleaning means not depicted can be provided so as to clean the scrubbing liquor prior to its entry into the absorption column 11.

The region of the lines 43, 13 may additionally be equipped with 40 (not depicted) feed lines for fresh scrubbing liquor in the event that the required inflow rate cannot be maintained with a regenerated scrubbing liquor alone.

The amount of inflowing scrubbing liquor can be controlled by 45 means of the pumps and by means of (not depicted) valve and throttle means.

The Examples which follow illustrate the invention.

Examples:

5 Example 1: Pilot plant for cleaning natural gas

A pilot plant for cleaning natural gas, consisting of an absorption column and a stripping column for regenerating the scrubbing liquor, was charged with 200 m<sup>3</sup>/h (s.t.p.) of natural gas as feed gas. At a pressure of 6 MPa (60 bar abs.) and a temperature of 40°C, the feed gas contained the following acidic constituents: 9.52%(v/v) of H<sub>2</sub>S, 2.99%(v/v) of CO<sub>2</sub> and 144 ppm<sub>v</sub> of CH<sub>3</sub>SH. The absorption region of the column was formed by a randomly dumped bed of 25 mm packing elements (IMTP 25, from 10 Norton) 9 m in depth. The absorber top of column temperature (i.e., the temperature of the regenerated scrubbing liquor passed in) was 40°C.

The gas was treated with 0.875 m<sup>3</sup>/h of a scrubbing solution which 20 had the following composition: 38.1% by weight of MDEA, 3.5% by weight of piperazine and 58.4% by weight of water. A virtually completely regenerated scrubbing liquor was used. The residual contamination of the scrubbing liquor was less than 2 m<sup>3</sup> of acid gases per 1000 kg of scrubbing liquor.

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The clean gas withdrawn at the top of the absorption column contained 0% of CO<sub>2</sub>, 5.3 ppm<sub>v</sub> of H<sub>2</sub>S, 26 ppm<sub>v</sub> of CH<sub>3</sub>SH. Thus, 84.2% of the CH<sub>3</sub>SH in the feed gas were removed.

30 Example 2: Commercial plant for cleaning natural gas

A commercial plant for cleaning natural gas, consisting of a 20 plate absorption column, a flash column and a stripping column, was charged with 53,625 m<sup>3</sup>/h (s.t.p.) of natural gas as feed gas. 35 At a pressure of 5.77 MPa (57.7 bar abs.) and a temperature of 30°C, the feed gas contained the following acidic constituents: 11.92%(v/v) of H<sub>2</sub>S, 5.86%(v/v) of CO<sub>2</sub>, 37 ppm<sub>v</sub> of CH<sub>3</sub>SH, 6.1 ppm<sub>v</sub> of C<sub>2</sub>H<sub>5</sub>SH and 3.3 ppm<sub>v</sub> of C<sub>3</sub>H<sub>7</sub>SH. The absorber top of column temperature was 48°C.

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The gas was treated with 180 m<sup>3</sup>/h of a scrubbing solution which had the following composition: 33.5% by weight of MDEA, 6.5% by weight of piperazine and 60.0% by weight of water. A virtually completely regenerated scrubbing liquor was used here as well.

45 The residual contamination of the scrubbing liquor was less than 2 m<sup>3</sup> of acid gases per 1000 kg of scrubbing liquor.

The clean gas contained 0.37% of  $\text{CO}_2$ , 3.0 ppm<sub>v</sub> of  $\text{H}_2\text{S}$ , 2.5 ppm<sub>v</sub> of  $\text{CH}_3\text{SH}$ , 1.0 ppm<sub>v</sub> of  $\text{C}_2\text{H}_5\text{SH}$  and 1.0 ppm<sub>v</sub> of  $\text{C}_3\text{H}_7\text{SH}$ . 94.4% of the  $\text{CH}_3\text{SH}$ , 86.6% of the  $\text{C}_2\text{H}_5\text{SH}$  and 75.6% of the  $\text{C}_3\text{H}_7\text{SH}$  in the feed gas were removed.

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We claim:

1. A process for removing mercaptans from a fluid stream comprising mercaptans and further acidic gases, especially CO<sub>2</sub> and/or H<sub>2</sub>S, which comprises intimately contacting the fluid stream in an absorption or extraction zone with a scrubbing liquor comprising at least one tertiary aliphatic alkanolamine of up to 12 carbon atoms and a primary or secondary amine in an amount of up to 20% by weight as activator, which is free of alkyl or dialkyl ethers of polyethylene glycol, the amount of scrubbing liquor being supplied to the absorption or extraction zone containing an excess of the aliphatic alkanolamine, based on the acidic gases to be removed, whereby at least CO<sub>2</sub> and H<sub>2</sub>S are removed essentially completely from the fluid stream, and separating the substantially decontaminated lean fluid stream and the loaden scrubbing liquor and discharging them from the absorption or extraction zone.
2. A process as claimed in claim 1, wherein the scrubbing liquor contains from 10 to 70% by weight of the alkanolamine.
3. A process as claimed in claim 1 or 2, wherein the scrubbing liquor contains not more than 5% by weight of a physical solvent for mercaptans.
4. A process as claimed in any of claims 1 to 3, wherein the tertiary alkanolamine used is methyldiethanolamine.
5. A process as claimed in any of claims 1 to 4, wherein the scrubbing liquor contains as activator, a primary or secondary alkanolamine or a saturated 5- or 6-membered N-heterocycle which optionally contains further heteroatoms selected from oxygen and nitrogen.
6. A process as claimed in claim 5, wherein the activator is selected from the group consisting of monoethanolamine, monomethylethanolamine, diethanolamine, piperazine, methylpiperazine and morpholine.
7. The use of a mixture of at least one aliphatic alkanolamine of 2 to 12 carbon atoms and at least one saturated 5- or 6-membered N-heterocycle which optionally contains further heteroatoms selected from O and N and which acts as an activator, for removing mercaptans from fluid streams

## Translation of amended sheet annexed to IPER

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comprising same, the mixture being free of monohydric and/or polyhydric alcohols.

8. The use as claimed in claim 7, wherein the alkanolamine is a 5 tertiary alkanolamine, preferably methyldiethanolamine.
9. The use as claimed in either of claims 7 and 8, wherein the activator is piperazine or methylpiperazine.
10. The use as claimed in claim 9, wherein the mixture contains from 10 to 70% by weight of methyldiethanolamine, from 0.5 to 15% by weight of piperazine, and water.

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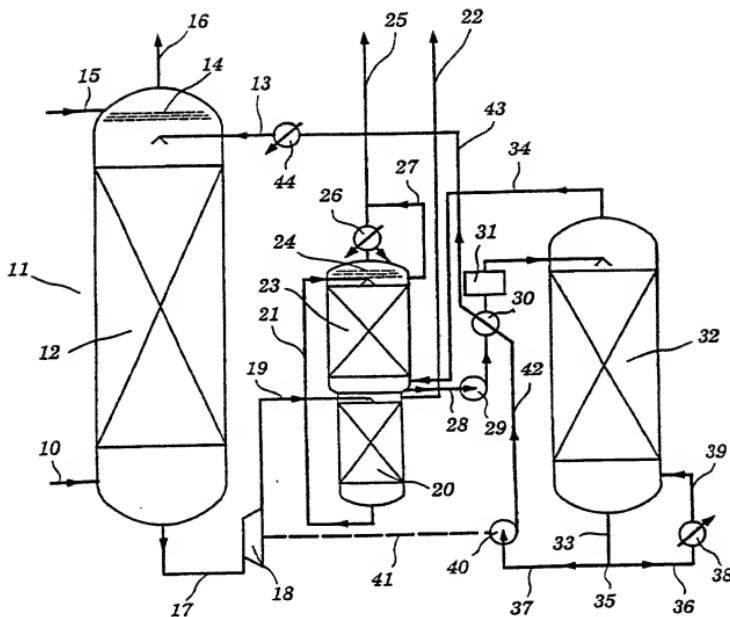
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# Declaration, Power of Attorney

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Page 1 of 3

0050/050486

We (I), the undersigned inventor(s), hereby declare(s) that:

My residence, post office address and citizenship are as stated below next to my name,

We (I) believe that we are (I am) the original, first, and joint (sole) inventor(s) of the subject matter which is claimed and for which a patent is sought on the invention entitled

Method for removing mercaptans from fluid fluxes

the specification of which

[ ] is attached hereto.

[ ] was filed on \_\_\_\_\_ as

Application Serial No. \_\_\_\_\_

and amended on \_\_\_\_\_.

[x] was filed as PCT international application

Number PCT/EPOO/ 06749

on July 14, 2000

and was amended under PCT Article 19

on \_\_\_\_\_ (if applicable).

We (I) hereby state that we (I) have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above.

We (I) acknowledge the duty to disclose information known to be material to the patentability of this application as defined in Section 1.56 of Title 37 Code of Federal Regulations.

We (I) hereby claim foreign priority benefits under 35 U.S.C. § 119(a)-(d) or § 365(b) of any foreign application(s) for patent or inventor's certificate, or § 365(a) of any PCT International application which designated at least one country other than the United States, listed below and have also identified below, by checking the box, any foreign application for patent or inventor's certificate, or PCT International application having a filing date before that of the application on which priority is claimed. Prior Foreign Application(s)

| Application No. | Country | Day/Month/Year | Priority Claimed |
|-----------------|---------|----------------|------------------|
| 19933301.7      | Germany | 15 July 1999   | [x] Yes [ ] No   |

We (I) hereby claim the benefit under Title 35, United States Codes, § 119(e) of any United States provisional application(s) listed below.

(Application Number)

(Filing Date)

(Application Number)

(Filing Date)

We (I) hereby claim the benefit under 35 U.S.C. § 120 of any United States application(s), or § 365(c) of any PCT International application designating the United States, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT International application in the manner provided by the first paragraph of 35 U.S.C. § 112, I acknowledge the duty to disclose information which is material to patentability as defined in 37 CFR § 1.56 which became available between the filing date of the prior application and the national or PCT International filing date of this application.

**Application Serial No.** **Filing Date** **Status (pending, patented, abandoned)**

And we (I) hereby appoint: Norman F. Oblon, Registration Number 24,618;  
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Robert W. Hahl, Registration Number 33,893, our (my) attorneys, with full  
powers of substitution and revocation, to prosecute this application and to transact all business in the Patent Office  
connected therewith; and we (I) hereby request that all correspondence regarding this application be sent to the firm of  
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We (I) declare that all statements made herein of our (my) own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

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